

Crystal data

C₁₅H₁₀O₂
M_r = 222.24
 Monoclinic
*P*2₁/*n*
a = 20.527 (2) Å
b = 6.144 (2) Å
c = 8.604 (2) Å
 β = 90.08 (2)°
V = 1085.0 (4) Å³
Z = 4
D_x = 1.360 Mg m⁻³
D_m not measured

Mo *K*α radiation
 λ = 0.71073 Å
 Cell parameters from 25 reflections
 θ = 13.4–15.0°
 μ = 0.084 mm⁻¹
T = 296 K
 Clear column
 0.35 × 0.19 × 0.19 mm
 Yellow

Data collection

AFC-5S diffractometer
 ω scans
 Absorption correction: none
 2788 measured reflections
 2717 independent reflections
 1377 reflections with
 $I > \sigma(I)$
R_{int} = 0.028
 θ_{\max} = 27.5°

h = 0 → 26
 k = 0 → 7
 l = -11 → 11
 6 standard reflections
 every 150 reflections
 intensity variation: ±1.8%
 (average maximum
 relative intensity)

Refinement

Refinement on *F*²
R = 0.054
wR = 0.053
S = 1.56
 1377 reflections
 195 parameters
 All H atoms refined
 $w = \sigma_F^{-2}$
 $(\Delta/\sigma)_{\max} < 0.01$
 $\Delta\rho_{\max} = 0.21 \text{ e } \text{Å}^{-3}$
 $\Delta\rho_{\min} = -0.20 \text{ e } \text{Å}^{-3}$

Extinction correction:
 Zachariasen (1963, 1968)
 Extinction coefficient:
 2.7 (3) × 10⁻⁶
 Scattering factors from
 Stewart, Davidson &
 Simpson (1965) for H,
 Cromer & Waber (1974)
 for C and O atoms

Table 1. Selected geometric parameters (Å, °)

O1—C15	1.315 (3)	C1—C15	1.485 (4)
O2—C15	1.219 (3)		
C2—C1—C15	118.0 (2)	O1—C15—C1	114.4 (3)
C13—C1—C15	122.1 (2)	O2—C15—C1	124.5 (2)
O1—C15—O2	121.1 (3)		

Table 2. Hydrogen-bonding geometry (Å, °)

D—H...A	D—H	H...A	D...A	D—H...A
O1—H1...O2 ⁱ	0.95 (4)	1.69 (4)	2.633 (3)	172 (3)

Symmetry code: (i) 1 - *x*, -1 - *y*, 1 - *z*.

Scan widths were (1.40 + 0.35tan θ)° in ω with a background/scan time-ratio of 0.5. The data were corrected for Lorentz and polarization effects. The Laue group assignment, systematic absences and intensity statistics consistent with centrosymmetry indicated space group *P*2₁/*n* (No. 14) and since refinement proceeded well it was adopted. Moreover, axial photos were consistent with monoclinic but not with orthorhombic symmetry. Fourier difference methods were used to locate the initial H-atom positions. The maximum effect of extinction was 9.7% of *F_o* for 21 $\bar{2}$. The maximum positive residual peak was located near the midpoint of the C1—C13 bond, while the maximum negative peak was located near the center of the C5—C8, C11, C12 ring.

Data collection: *MSCIAFC Diffractometer Control Software* (Molecular Structure Corporation, 1988). Cell refinement: *MSCIAFC Diffractometer Control Software*. Data reduction: *TEXSAN* (Molecular Structure Corporation, 1989). Program(s) used to solve structure: *SHELXS86* (Sheldrick, 1985). Program(s) used to refine structure: *TEXSAN*. Molecular graphics: *ORTEPII* (Johnson, 1976). Software used to prepare material for publication: *TEXSAN*.

Support provided to LJF by PPG Industries is gratefully acknowledged. We thank Dr J. C. Gallucci for help of various sorts. The diffractometer was purchased with funds provided in part by an NIH grant.

Supplementary data for this paper are available from the IUCr electronic archives (Reference: BK1321). Services for accessing these data are described at the back of the journal.

References

- Bondi, A. (1964). *J. Phys. Chem.* **68**, 441–451.
 Cromer, D. T. & Waber, J. T. (1974). *International Tables for X-ray Crystallography*, Vol. IV, pp. 71, 148. Birmingham: Kynoch Press. (Present distributor Kluwer Academic Publishers, Dordrecht.)
 Fitzgerald, L. J. & Gerkin, R. E. (1993). *Acta Cryst.* **C49**, 1952–1958.
 Fitzgerald, L. J. & Gerkin, R. E. (1996). *Acta Cryst.* **C52**, 1838–1841.
 Fitzgerald, L. J. & Gerkin, R. E. (1997). *Acta Cryst.* **C53**, 71–73.
 Johnson, C. K. (1976). *ORTEPII*. Report ORNL-5138. Oak Ridge National Laboratory, Tennessee, USA.
 Molecular Structure Corporation (1988). *MSCIAFC Diffractometer Control Software*. MSC, 3200 Research Forest Drive, The Woodlands, TX 77381, USA.
 Molecular Structure Corporation (1989). *TEXSAN. TEXRAY Structure Analysis Software*. MSC, 3200 Research Forest Drive, The Woodlands, TX 77381, USA.
 Sheldrick, G. M. (1985). *SHELXS86. Crystallographic Computing 3*, edited by G. M. Sheldrick, C. Krüger & R. Goddard, pp. 175–189. Oxford University Press.
 Stewart, R. F., Davidson, E. R. & Simpson, W. T. (1965). *J. Chem. Phys.* **42**, 3175–3187.
 Zachariasen, W. H. (1963). *Acta Cryst.* **16**, 1139–1144.
 Zachariasen, W. H. (1968). *Acta Cryst.* **A24**, 212–216.

Acta Cryst. (1997). **C53**, 1082–1084

An Estrone–Glucuronide Conjugate

C. MARK SMALES, LEONARD F. BLACKWELL, JOYCE M. WATERS AND ANTHONY K. BURRELL

Department of Chemistry, Massey University, Private Bag 11222, Palmerston North, New Zealand. E-mail: a.k.burrell@massey.ac.nz

(Received 14 October 1996; accepted 20 March 1997)

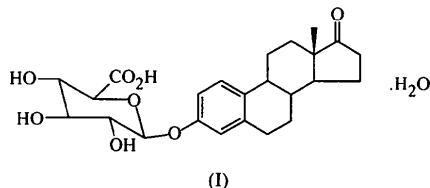
Abstract

The crystal structure of 2-*O*-[17-oxoestra-1,3,5(10)-trien-3-yl]- β -D-glucopyranosiduronic acid hydrate, C₂₄H₃₀O₈·

H₂O, is reported. The angle between the mean plane of the A ring of the steroid and the mean plane of the glucuronide is 14.1 (5)°.

Comment

The X-ray structure determination of the title compound, (I), was performed as part of a program to characterize structurally the lysozyme–estrone glucuronide conjugates used in the Ovarian Monitor homogeneous enzyme immunoassay for the delineation of the fertile period (Brown, Blackwell, Holmes & Smyth, 1989). It is necessary to have the structural coordinates of (I) so that an accurate three-dimensional picture of the lysozyme conjugates may be generated by computer modeling based on the known crystal structure of hen egg-white lysozyme. The coordinates of the steroid glucuronide moiety are required so that the computer model correctly represents the possible orientations of the steroid glucuronide with respect to the lysozyme surface. The sites of attachment of the steroid glucuronide in the conjugates have been determined (Smales, 1997); hence the spatial requirements of the antibody–conjugate interaction can now be investigated for the different estrone glucuronide–lysozyme conjugates by computer graphic techniques.



Although the crystal structures of many steroids have been published (for example see Duax, Weeks, & Rohrer, 1976) and the structure of estrone has been known for some time (Busetta, Courseille & Hospital, 1973), there are very few examples of the crystal structures of steroid glucuronides. Both our recently published estriol 17 β -glucuronide structure (Wu, Waters & Blackwell, 1996) and the reported estradiol 17 β -glucuronide structure of Hadd *et al.* (1983) contained sugar residues which were acetylated at the sugar hydroxyl groups. Also, in these compounds, the glycosidic linkage between the steroid and the glucuronide ring was *via* the C17 position of the steroid skeleton D ring. So far, there are no reported crystal structures of steroid glucuronides where the glycosidic linkage is located at the C3 position of the steroid A ring and the hydroxyl groups and the carboxyl group of the carbohydrate moiety are free.

The title compound crystallizes along with a single water molecule which appears to form a hydrogen-bonding interaction between the O6 hydroxyl group on one steroid molecule and the O7 hydroxyl group on a second steroid molecule. All bond lengths and angles

in the steroid skeleton are in the expected ranges. The H atom associated with the carboxylic acid group and those of the solvent water molecule could not be located and are omitted. The angle between the mean plane of the A ring of the steroid (C1–C5, C10) and the mean plane of the glucuronide (C19–C23, O3) is 14.1 (5)°, with a torsion angle (C3, O2, C19, C20) of 158.4 (3)°. The structure also shows that the carbohydrate ring is in a chair conformation and that all of the H-atom substituents of the glucuronide ring occupy axial sites; this results in the adoption of *trans* positions for the H atoms attached to adjacent C atoms with respect to each other. Hence, the orientation of the critical linkage between the carbohydrate moiety and the C3 position of the steroid A ring, which determines the orientation of the steroid upon conjugation to lysozyme, is clearly shown.

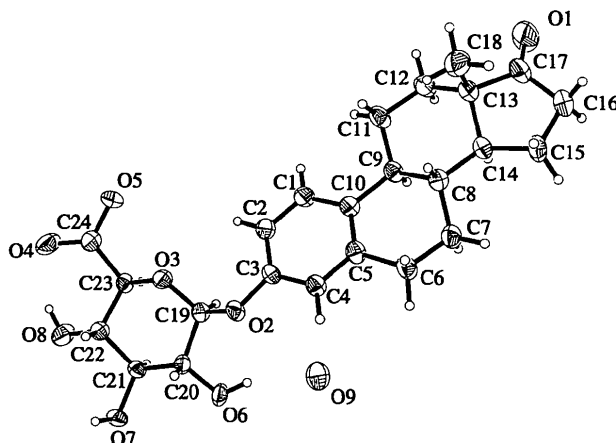


Fig. 1. A ZORTEP (Zsolnai, 1994b) drawing of (I) with displacement ellipsoids drawn at the 50% probability level.

Experimental

The title compound was synthesized *via* a Koenigs–Knorr (Conrow & Bernstein, 1971) coupling reaction as previously described by Smales, Cooke & Blackwell (1994). The compound was crystallized from a mixture of methanol and Milli-Q water at room temperature.

Crystal data

C₂₄H₃₀O₈·H₂O
M_r = 464.50
 Orthorhombic
*P*2₁2₁2₁
a = 7.0337 (2) Å
b = 8.3981 (2) Å
c = 37.9873 (11) Å
V = 2243.90 (11) Å³
Z = 4
D_x = 1.375 Mg m⁻³
D_m not measured

Mo *K*α radiation
 λ = 0.71073 Å
 Cell parameters from 6916 reflections
 θ = 1–25°
 μ = 0.105 mm⁻¹
T = 293 (2) K
 Plate
 0.500 × 0.120 × 0.035 mm
 Colorless

Data collection

Siemens SMART diffractometer	$R_{\text{int}} = 0.046$
Absorption correction: none	$\theta_{\text{max}} = 23^\circ$
10 397 measured reflections	$h = -8 \rightarrow 8$
3111 independent reflections	$k = -8 \rightarrow 10$
2730 reflections with $I > 2\sigma(I)$	$l = -44 \rightarrow 47$
	Intensity decay: $< 2\%$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0479P)^2 + 1.3262P]$
$R[F^2 > 2\sigma(F^2)] = 0.051$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.133$	$(\Delta/\sigma)_{\text{max}} = -0.066$
$S = 1.132$	$\Delta\rho_{\text{max}} = 0.272 \text{ e } \text{\AA}^{-3}$
3110 reflections	$\Delta\rho_{\text{min}} = -0.259 \text{ e } \text{\AA}^{-3}$
337 parameters	Extinction correction: none
Only coordinates of H atoms refined	Scattering factors from <i>International Tables for Crystallography</i> (Vol. C)

Data collection: XSCANS (Siemens, 1994). Cell refinement: XSCANS. Data reduction: SAINT (Siemens, 1995). Program(s) used to solve structure: SHELXS86 (Sheldrick, 1985). Program(s) used to refine structure: SHELXL93 (Sheldrick, 1993). Molecular graphics: XPMA (Zsolnai, 1994a) and ZORTEP (Zsolnai, 1994b). Software used to prepare material for publication: SHELXL93.

The authors thank Associate Professor C. E. F. Rickard (University of Auckland) for data collection.

Supplementary data for this paper are available from the IUCr electronic archives (Reference: TA1138). Services for accessing these data are described at the back of the journal.

References

- Brown, J. B., Blackwell, L. F., Holmes, J. & Smyth, K. (1989). *Int. J. Gynaecol. Obstet. Suppl.* **1**, 111–122.
- Busetta, P. B., Courseille, C. & Hospital, M. (1973). *Acta Cryst.* **B29**, 298–313.
- Conrow, R. B. & Bernstein, S. (1971). *J. Org. Chem.* **36**, 863–870.
- Duax, W. L., Weeks, C. M. & Rohrer, D. C. (1976). *Topics in Stereochemistry*, Vol. 9, edited by E. L. Eliel & N. Allinger, pp. 271–383. New York: John Wiley.
- Hadd, H. E., Slikker, W., Miller, D. W., Helton, E. D., Duax, W. L., Strong, P. D. & Swenson, D. C. (1983). *J. Steroid Biochem.* **18**, 81–87.
- Smales, C. M. (1997). PhD thesis, Massey University, New Zealand.
- Smales, C. M., Cooke, D. G. & Blackwell, L. F. (1994). *J. Chromatogr. B*, **662**, 3–14.
- Sheldrick, G. M. (1985). *SHELXS86. Program for the Solution of Crystal Structures*. University of Göttingen, Germany.
- Sheldrick, G. M. (1993). *SHELXL93. Program for the Refinement of Crystal Structures*. University of Göttingen, Germany.
- Siemens (1994). *XSCANS. X-ray Single Crystal Analysis System*. Version 2.1. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
- Siemens (1995). *ASTRO and SAINT. Data Collection and Processing Software for the SMART System*. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
- Wu, Y., Waters, J. M. & Blackwell, L. F. (1996). *J. Chem. Soc. Perkin Trans. 2*, pp. 1449–1453.

Zsolnai, L. (1994a). *XPMA. Program for Molecular Graphics*. University of Heidelberg, Germany.

Zsolnai, L. (1994b). *ZORTEP. Interactive Graphics Program*. University of Heidelberg, Germany.

Acta Cryst. (1997). **C53**, 1084–1086

endo-3-Trimethylsilyl-2-norbornyl 2-Nitrobenzenesulfinate

JONATHAN M. WHITE AND ALISON J. GREEN

School of Chemistry, University of Melbourne, Parkville, Victoria 3052, Australia. E-mail: jonathan.white@muwayf.unimelb.edu.au

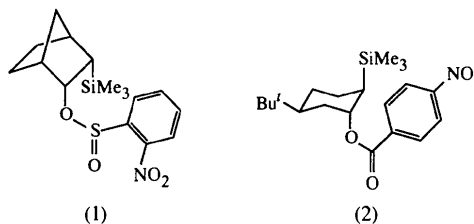
(Received 17 October 1996; accepted 3 March 1997)

Abstract

The structure of the title compound, C₁₆H₂₃NO₄SSi, was solved in order to determine the C(alkyl)—O(ester) bond distance for comparison purposes. There exists a close contact [O3...S 2.766(2) Å] between one of the nitro O atoms and the sulfinate S atom. The C—O bond distance is 1.464(3) Å.

Comment

As part of our studies of the effects of silicon substituents on C—O bond lengths at the β position (White & Robertson, 1992; Kuan, Green & White, 1995; Chan *et al.*, 1996), we carried out a structural study on the title compound, (1). The norbornyl framework of (1) which was expected to enforce a synperiplanar relationship between the trimethylsilyl substituent and the 2-nitrobenzenesulfinate substituent would allow us to investigate the effects of the $\sigma(\text{C—Si})-\sigma^*(\text{C—O})$ interaction in this geometry. The $\sigma(\text{C—Si})-\sigma^*(\text{C—O})$ interaction in the antiperiplanar geometry, as in (2), has been shown to lead to significant lengthening of the C—O bond length (White & Robertson, 1992; Kuan, Green & White, 1995).



Compound (1) was prepared from *endo*-3-trimethylsilyl-*endo*-2-norborneol, (3) (Lambert & Chelius, 1990),